

8-25-14 AP Chem

Classify each of the substances as being soluble or insoluble in water.

- a. $\text{KBr} =$
- b. $\text{PbCO}_3 =$
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- d. zinc hydroxide =
- e. sodium acetate =
- f. silver iodide =
- g. cadmium (II) sulfide =
- h. zinc carbonate =
- i. $\text{FeCl}_3 =$
- j. $\text{KCl} =$
- k. $(\text{NH}_4)_2\text{SO}_4 =$
- l. $\text{Mn}(\text{OH})_2 =$
- m. $\text{AgCl} =$
- n. $\text{BaSO}_4 =$

Then either name or write the formula!