### Purple (and Red, Orange, Yellow) Power

# Your mom might tell you, the lunch ladies might tell you, your doctor might tell you ... eat those colorful fruits and vegetables! But why?

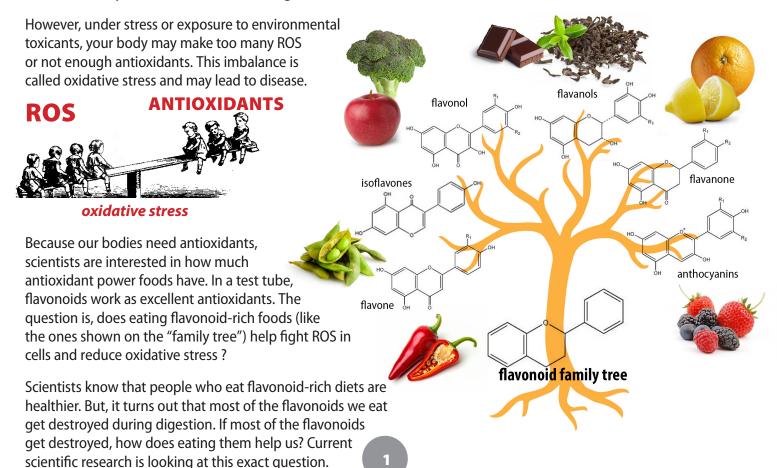
Most of the color in foods comes from a big family of chemicals called *flavonoids*. The name comes from the Latin word *flavus* which means yellow. If your diet is rich in fruits and vegetables, you consume a lot of flavonoids. That's a good thing, since scientists are learning more all the time about how flavonoids improve health and help reduce the risk of heart disease and certain cancers.

Scientists divide the flavonoid family into six smaller groups. You can see the names of these groups and some of the foods from each group on the "family tree" below. Take a look at the chemical structures for all six flavonoid groups. What do you notice that is similar about all of them? What variations do you see that are unique to each group?



#### The power of antioxidants

Your cells constantly obtain food and convert it to energy. This chemical process is called cell metabolism. Cell metabolism is essential to life, but cell metabolism also produces chemicals that can damage cells called *reactive oxygen species* (ROS). Your body creates a constant supply of another group of chemicals called *antioxidants* to react with the ROS before they can cause too much damage.



#### One flavonoid research story

Scientists like <u>Bernie Hennig</u> at the University of Kentucky are doing exciting new research on how flavonoids can help protect our cells. New research is showing that certain flavonoids (from the *flavonol* and *flavanol* groups) can protect cells from damage caused by environmental toxicants called PCBs.

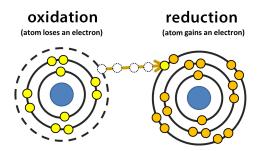
PCBs can turn on certain genes which then produce enzymes. These enzymes cause inflammation of the cells that line blood vessels, leading to a disease called atherosclerosis (a buildup of fats, cholesterol and other substances in and on artery walls). The flavonols and flavanols slow down production of the damaging enzymes and also reduce the harmful effects of any enzymes that are still produced.

## Why do plants have flavonoids?

We might enjoy and benefit from eating flavonoid-rich foods, but plants don't make flavonoids in their cells for us! Plants invest their energy and nutrients in making flavonoids, and the plants get a payback for that! This diverse group of chemicals benefits plants in lots of ways. The colors of flowers attract pollinators such as bees and butterflies; the colors of fruits attract animals who eat the fruit and disperse the seeds. Flavonoids help seeds germinate and help seedlings grow and develop. Flavonoids also protect plants from frost and drought and filter out ultra-violet (UV) light to protect tissues.

## What are oxidation and reduction?

The ROS that can damage cells react through a process called *oxidation*. Oxidation is the loss of electrons by a molecule, atom, or ion. Oxidation is always paired with reduction. *Reduction* is the gain of electrons by a molecule, atom, or ion.



Protecting cells from ROES damage is important. Antioxidants are chemicals that react with ROS before they can react with other cell chemicals in a harmful manner. Plant and animal cells contain many types of antioxidants. For example, vitamins A, C, and E are antioxidants. Cells manufacture antioxidants and we also get antioxidants from food.

#### Anthocyanins—flavonoids with color tricks



The cups in the picture all contain juice from purple cabbage. The only difference between them is the pH of the liquid.

Anthocyanins are the type of flavonoids found in blue, purple, and red foods such as blueberries, cranberries, blackberries, cherries, grapes, purple cabbage, and eggplant. They are UV filters in plants.

Anthocyanins are especially interesting because they change color depending on acidity. You may remember that household items like lemon juice and vinegar are acids and baking soda and ammonia are bases. Scientists represent how acidic or basic something is using a scale called pH, where a pH of less than 7 is

acidic, a pH of more than 7 is basic, and a pH of 7 is neutral.

Anthocyanins can be used as pH indicators because they are pink in acidic solutions, purple in neutral solutions, and greenish-yellow in basic solutions.

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