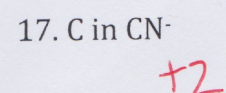
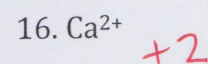
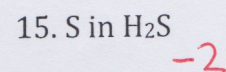
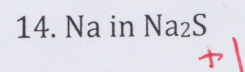
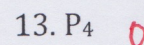
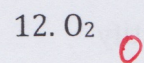
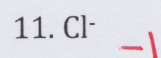
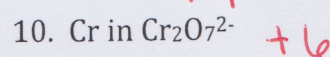
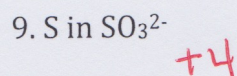
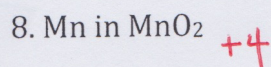
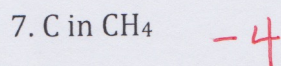


## Oxidation Numbers

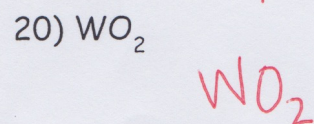
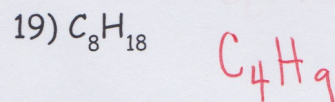
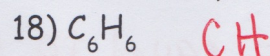
You must know the rules and be able to use them to assign oxidation numbers to any atom.



## Empirical and Molecular formula calculations

Be able to use % composition to determine the formula of a compound

Write the empirical formula for the following compounds.



Just  
reduce  
subscripts